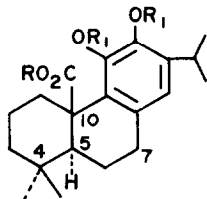


STEREOCHEMISTRY OF SALVIN AND
PICROSALVIN

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THE structure of the new diterpene (I) which we would now identify by the trivial name 'salvin', has been well-established (1), leaving its stereochemistry uncertain. We would now like to propose the A/B rings to be trans-fused in the compound as shown in (I). PMR spectra in pyridine solution show that one of the methyl groups at C₄ is deshielded by 19 cps in the acid (II) compared with the

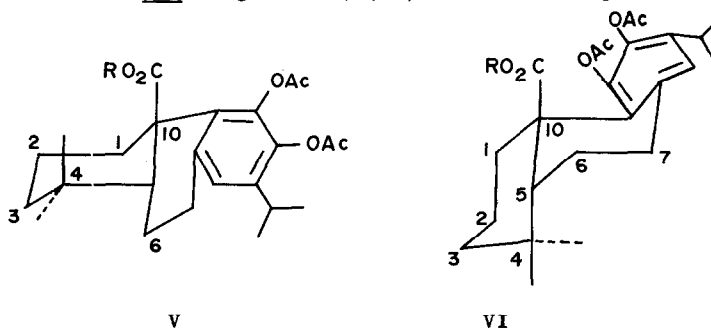


- I R = R₁ = H
II R = H; R₁ = Ac
III R = CH₃; R₁ = Ac
IV R = R₁ = CH₃

same signal in the ester (III). This clearly indicates that one of the C₄ methyl groups is in a 1,3-cis diaxial relationship with the carboxyl group at C₁₀ (2). This can happen only if (a) the A/B rings are trans fused as in (II) or (b) they are cis and assume conformation (V) in preference to (VI). From the non-bonded interactions enumerated (which are not common to both) that destabilise each of the conformations it may be safely assumed (3)

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that with a cis ring fusion, (VI) would be the preferred

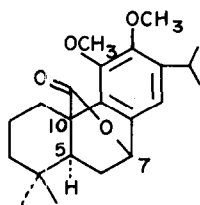


Non-bonded interactions that destabilise V and VI.

- | V | VI |
|---|--|
| (1) C ₁₀ -CO ₂ R, C ₄ β-Me | (1) C ₁₀ -CO ₂ R, C ₆ β-H |
| (2) C ₁₀ -CO ₂ R, C ₂ β-H | (2) C ₄ α-CH ₃ , C ₇ α-H |
| (3) C ₆ -CH ₂ , C ₁ α-H | |
| (4) C ₆ -CH ₂ , C ₃ α-H | |

conformation of the molecule. This, however, would not show any 1,3-diaxial relationship with a C₄-methyl and a C₁₀-carboxyl group. Hence the A/B rings should be trans-fused.

Picrosalvin has been shown to have structure (VII)



(VII)

and is synthesized from salvin (I) by oxidising (IV) to a C₇ ketone, reducing it with NaBH₄ to a C₇-hydroxyl derivative, and then lactonising (1). Hence both the diterpenes have the same stereochemistry at C₅ and C₁₀. The specific rotations of the acid (II) [α]_D +140°, of the

lactone (VII), $[\alpha]_D -73.5^\circ$ and that of the corresponding hydroxy acid obtained by hydrolysing the lactone (VII), $[\alpha]_D +31^\circ$ clearly indicate that (I) and (VII) represent the absolute stereochemistry of the two compounds (4).

Independently, E.Wenkert et al. have also recently come to the same conclusion by other methods (5), wherein they have renamed picrosalvin as carnosol.

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